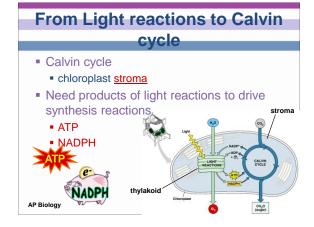
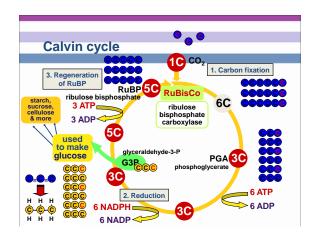
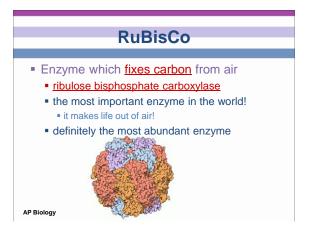


From $CO_2 \rightarrow C_6H_{12}O_6$

- CO₂ has very little chemical energy
 - fully oxidized
- C₆H₁₂O₆ contains a lot of chemical energy
 - highly reduced
- Synthesis = endergonic process
 - put in a lot of energy
- Reduction of CO₂ → C₆H₁₂O₆ proceeds in many small uphill steps
 - each catalyzed by a specific enzyme
- using energy stored in ATP & NADPH P Biology







■ The accounting is complicated ■ 3 turns of Calvin cycle = 1 G3P ■ $3 \text{ CO}_2 \rightarrow 1 \text{ G3P}$ (3C) ■ $6 \text{ turns of Calvin cycle} = 1 \text{ C}_6 \text{H}_{12} \text{O}_6$ (6C) ■ $6 \text{ CO}_2 \rightarrow 1 \text{ C}_6 \text{H}_{12} \text{O}_6$ (6C) ■ $18 \text{ ATP} + 12 \text{ NADPH} \rightarrow 1 \text{ C}_6 \text{H}_{12} \text{O}_6$ ■ any ATP left over from light reactions will be AP Biology Used elsewhere by the cell

